

Fabrication and Characterisation of Titanium Dioxide (TiO₂)/PVA Nanofibre Composites using Electrospinning

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Abstract

A study on the formation of titanium dioxide (TiO₂)/ polyvinyl alcohol (PVA) nanofibres has been investigated in the research work. In the study, the TiO₂/PVA solutions were prepared at different concentrations ranging from 1wt% to 10wt%. The PVA solution without TiO₂ was used as control sample in the study. Each solution was extruded using electrospinning at different voltages to form long and continuous nanofibres. The fibres were then characterised for morphological structures, fibre diameter and membrane area. The result shows that control sample of PVA nanofibres formed beads on the fibre surfaces. The addition of TiO₂ in PVA was found to reduce the bead formation. The clumps of TiO₂ nanoparticles were observed and were also confirmed with the EDX mapping. In addition, the electrospinning parameters affect the formation of TiO₂/PVA nanofibers. At 15 kV, the resultant fibre diameter increases from 153 ± 23 nm to 191 ± 26 nm when the TiO₂ concentration was further increased from 1wt% to 10wt%. The morphological structure of TiO₂/PVA nanofibres varies depending on applied voltages and concentrations used. The membrane area of TiO₂/PVA nanofibers is approximately 187 cm² for 1wt% of TiO₂. From the study, it shows that an optimal concentration to produce TiO₂/ PVA nanofibers is 10wt%. At 10wt%, no beads were observed on the fibre and the TiO₂ nano-particles were uniformly distributed on the membrane surfaces.

Keywords: Titanium dioxide (TiO₂) nano-particles; polyvinyl alcohol (PVA); electrospinning; morphological structures; concentration..

1. Introduction

Titanium dioxide (TiO₂) is a powder-based chemical which is low cost, chemically stable and non-toxic [1]. In the past, titanium dioxide has been studied extensively in a broad range of industrial and scientific field [2]. Titanium dioxide (TiO₂) has many interesting properties such as high photo activity, self-cleaning and anti-microbial which leads to receive much attention from researchers [3]. TiO₂ used to be applied on different substrates such as activated carbon, glass, stainless steel [4] and silica [5]. However, the use of TiO₂ for conventional separation method was rather difficult and it becomes the drawback for the wide application of TiO₂ [6]. In this study, the TiO₂ nanoparticles have been bounded into composite nanofibres using polyvinyl alcohol (PVA) as fibre template throughout electrospinning process.

Electrospinning is a facile and straightforward process that uses a high-voltage electric field to produce ultrafine nanofibres [7]. Electrospinning is basically operated by using the concept of electrostatic that impart towards polymer solution in order to produce fine fibres with the range of 100nm-500nm. According Mirjalili *et al.* [8], electrospinning occurs when the polymer solution is subjected to certain heat and forced to form droplet at the tip of syringe. Then, high voltage applied to polymer solution to aid the polymer acceleration on the surface of collector in order to form nanofibers. The accumulation of the nanofibre forms a porous membrane.

Polyvinyl Alcohol (PVA) is one of synthetic polymers that can be spun using electrospinning. PVA nanofibres produced from the electrospinning technique have gained a great deal of attention due

to their unique contribution of properties such as highly hydrophilic, small pore sizes, fine fibre diameter, light weight and high surface area [9-12]. Due to these properties, the PVA nanofibres have potential to be employed in various applications such as filtration [8-9] and biomaterials [9-12]. In addition, the PVA can be incorporated with any inorganic materials such as titanium dioxide (TiO₂), carbon nanotube (CNT) etc. Therefore, the aim of the study is to investigate the spinnability of TiO₂/PVA solution using electrospinning. This research study focuses on identifying an ideal concentration of titanium dioxide (TiO₂) and polyvinyl alcohol (PVA) for the formation of nanofibre composite as well as characterizing fibre morphology, fibre diameter and membrane area for titanium dioxide (TiO₂)/polyvinyl Alcohol (PVA) nanofibre composite.

2. Materials and method

2.1. Materials

Polyvinyl alcohol (PVA) (Sigma-Aldrich, Mw 125,000 g/mol) and titanium dioxide (TiO₂) nano-particles (Sigma-Aldrich) were used as received. Distilled water was used as solvent in the experiments.

2.2. Preparation of control PVA and TiO₂/PVA solutions

PVA and TiO₂ nano-particles were dissolved in distilled water at 120°C for 4 hours. The concentration of TiO₂ was in a range of 1wt% to 10 wt%. The solutions were left for 24 hours at room

temperature electrospinning process. The preparation of control PVA solution was briefly described elsewhere [9].

2.3. Electrospinning process

Fig. 1 illustrates the fabrication of control PVA and TiO₂/PVA nanofibre composites using electrospinning. In this study, the control PVA and TiO₂/PVA solutions were electrospun at constant distance of 10cm with 0.5ml/hr of flow rate. Each solution was electrospun at different applied voltages ranging from 10kV to 15k. The control PVA and TiO₂/PVA nanofibres composites were collected onto grounded aluminium foil for about 5 minutes. The resultant fibres and membranes were characterized for morphological structure, fibre diameter and membrane area.

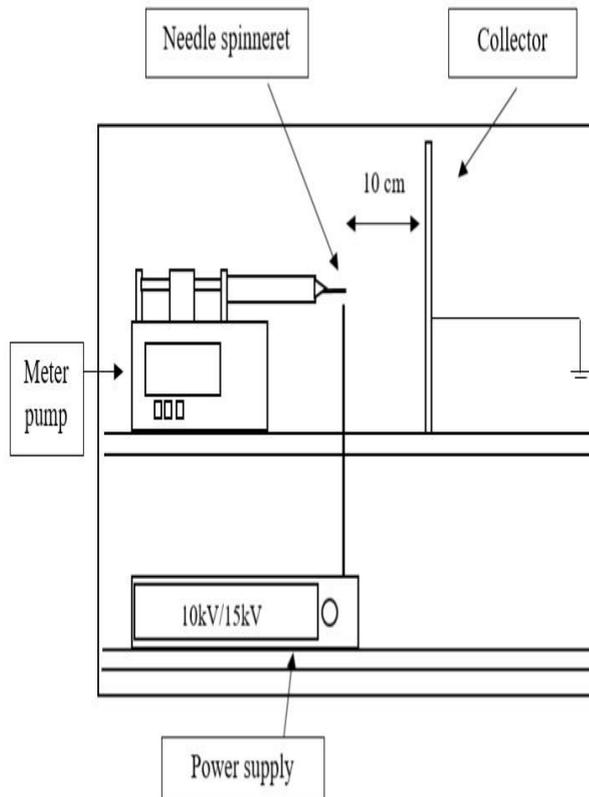


Fig. 1: Schematic diagram of electrospinning process.

2.4. Fibre and membrane characterization

For fibre diameter, 50 measurements were taken at random places and were measured using ImageJ software.

For membrane area, the fabricated membrane was scanned over a transparent grid with an area of 1cm² per grid. The scanned image was then measured using ImageJ software. Three measurements were taken for each sample.

The presence of TiO₂ nano-particle on fibre surface was tested using EDX mapping.

3. Results and discussion

3.1. Effects of TiO₂ concentrations on the morphological structure of TiO₂/PVA nanofibres composites.

Figs. 2 (a)-(d) illustrate the formation of control PVA (polyvinyl alcohol) and TiO₂/PVA nanofibre composites using electrospinning. From the results, it shows that control PVA and 1wt% TiO₂/PVA nanofibre composites have long and continuous fibres with beads on the fibre surfaces (Figs. 2 (a)-(b)).

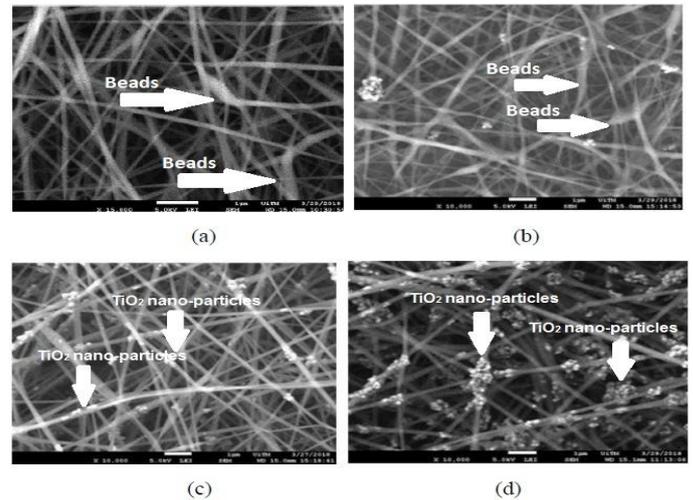


Fig. 2: Typical SEM images of (a) control sample PVA, (b) 1wt% TiO₂/PVA, (c) 5wt% TiO₂/PVA and (d) 10wt% TiO₂/PVA, respectively.

The formation of beads was probably due to low viscosity of solution for control PVA and 1wt% TiO₂/PVA. At lower viscosity, the solution was insufficient to overcome high columbic repulsion force during electrospinning process, which resulted in beads on fibre surfaces. Similar finding was observed by Zhang *et.al* [10], where they reported that solution with low viscosity produced beaded fibres.

As the concentration increases from 5wt% to 10wt% the amount of beads reduce (Figs. 2 (c)-(d)). This indicates that solution with high viscosity has sufficient polymer chain entanglement that able to withstand high columbic repulsion force during electrospinning. The result was in agreement with Zhang *et.al* [10]. They observed the transition from beaded fibres to uniform cylindrical fibres as the PVA viscosity increased [10].

In addition, the TiO₂ nanoparticles are well distributed as the concentration increases from 5wt% to 10wt%. This also shows that solution with high concentration of TiO₂ (10wt%) disperses more TiO₂ nanoparticles on fibre surfaces than solutions at 1wt% and 5wt%.

The presence of the TiO₂ nanoparticles was analyzed using EDX mapping. As shown in Fig. 3, the Ti peaks prove the existence of TiO₂ nanoparticles on the fibre surfaces. Al peaks indicates backing material of the nanofibres.

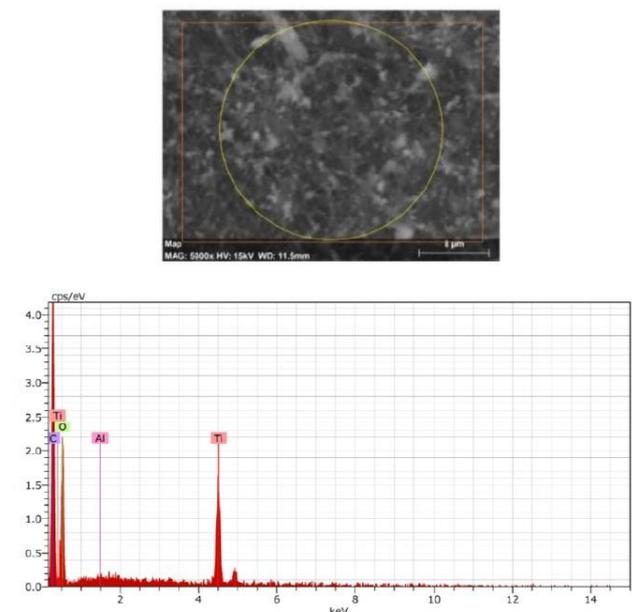


Fig. 3: Typical EDX mapping for TiO₂/PVA nanofibres.

3.2. Effect of applied voltage on morphological structure and fibres diameter of control sample PVA and TiO₂/PVA nanofibres.

Figs. 4 (a)-(h) depict the effect of applied voltages on morphological structures of control PVA and TiO₂/PVA nanofibre composites. The increase of applied voltages produced beaded fibres for control PVA and 1wt% TiO₂/PVA. As mentioned earlier, the formation of beaded fibres was due to the low solution viscosity where the solution was insufficient to overcome high columbic repulsion force during electrospinning process.

For 5wt% and 10wt%, cylindrical fibres with TiO₂ nano-particles were fabricated at any voltages. This indicates that solutions at 5wt% TiO₂ and 10wt% TiO₂ have sufficient polymer chain entanglement that able to withstand high columbic repulsion force during electrospinning. As reported Zhang *et.al* [10], the transition from beaded fibres to uniform cylindrical was due to the increase of PVA solution concentration [10].

As shown in Fig.5, the increment of voltages from 10kV to 15kV reduced the control PVA fibre diameter from 178 + 29 nm to 147 + 23 nm. It was expected that an increase in applied voltage gave higher elongation to the polymer jet which resulted in the reduction of fibre diameter. The reduction of fibre diameter at higher voltage was also observed by other recent studies [9-11]

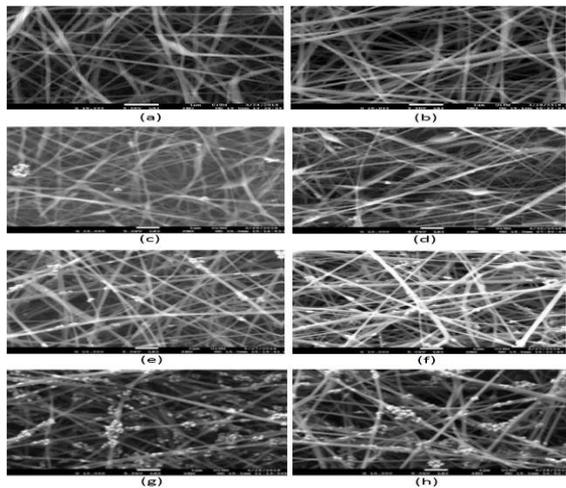


Fig. 4: Typical SEM images of control sample PVA at (a)10kV and (b) 15kV, 1wt% TiO₂/PVA nanofibres at (c) 10kV and (d) 15kV, 5wt% TiO₂/PVA nanofibres at (e) 10kV and (f) 15kV and 10wt% TiO₂/PVA nanofibres at (g) 10kV and (h) 15kV.

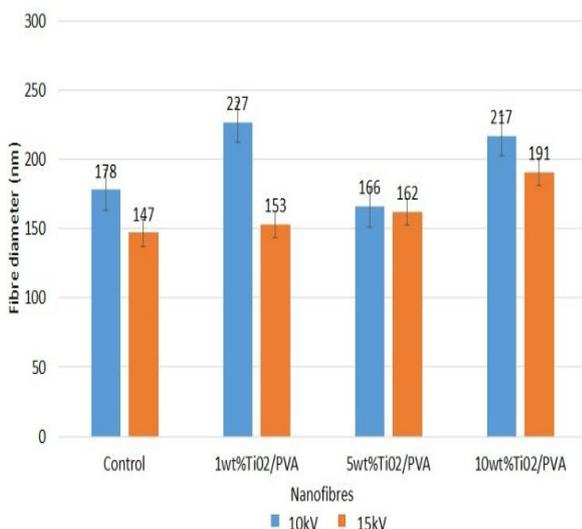


Fig. 5: Average fibre diameter against control sample PVA and TiO₂/PVA nanofibre composites at 10kV and 15kV.

For 1wt%TiO₂/PVA nanofibres, it was found that the diameter decreased from 227 + 29 nm to 153 + 23 nm with increasing of applied voltage. Similar result was also found for 5wt%TiO₂/PVA nanofibre composites, where the fibre diameter reduced from 166 + 29 nm to 162 + 23 nm with increasing of applied voltage from 10kV to 15kV. The reduction of fibre diameter was also found for 10wt%TiO₂/PVA. As shown in Fig.5, the fibre diameter of 10wt%TiO₂/PVA nanofibre composites reduces from 217 + 23nm to 191 + 28nm when the voltages decrease from 10kV to 15kV. Recent studies reported that the reduction of fibre diameter was due to the further elongation of fibre at higher voltages [9-11]. In addition, Khajavi and Damerchely [12] reported the resultant PVA nanofibres diameter was dependent on voltage and concentration used.

3.3. Effects of applied voltage on membrane area

Table 1 tabulates membrane area for control PVA and TiO₂/PVA nanofibre composite. The membrane area of control PVA was found to increase from 127 ± 1.5 cm² to 185 ± 1.3 cm² with increasing of voltages. When a higher voltage was applied, jets with bigger bending instabilities could form. As a result, a large membrane area was deposited onto collector. The results were in agreement with Affandi NDN [11], where the author observed the increase of membrane area at higher voltage.

However, for TiO₂/PVA membrane, the membrane area decreases when the voltages increase from 10kV to 15kV. As shown in Table 1, the membrane area of 1wt% TiO₂/PVA nanofibre composites reduces from 187 ± 1.2 cm² to 123 ± 3.0 cm². The membrane area for 5wt% TiO₂/PVA and 10wt% TiO₂/PVA also reduced with increasing of applied voltage ranging from 10kV to 15kV. It was expected that the presence of TiO₂ in solution produced unstable charged jet at higher voltage. This has resulted in a small membrane area for TiO₂/PVA nanofibre composites. Similar finding also observed by Supaphol *et. al* [13], where the authors reported that an increase in applied voltage reduced the area of electrospun nanofibre membrane.

Table 1: Area of control PVA and TiO₂/PVA nanofibre membranes

Applied voltage (kV)	Area of membrane (cm ²)			
	Control sample PVA	1wt% TiO ₂ /PVA	5wt% TiO ₂ /PVA	10wt% TiO ₂ /PVA
10	127 + 1.5	187 + 1.2	180 + 1.4	186 + 1.5
15	185 + 1.3	123 + 3.0	177 + 2.6	169 + 1.2

4. Conclusion

The incorporation between PVA and titanium dioxide were investigated and the optimal TiO₂/PVA concentration was identified in the study. The result shows that the optimal concentration for TiO₂/PVA solution was 10wt%. At 10wt%, no beads were observed on the fibre and the TiO₂ nano-particles were uniformly distributed on the membrane surfaces. The solution concentration also affects morphological structures of TiO₂/PVA nanofibre composites. The control sample of PVA nanofibres formed beads on the fibre, whereas the TiO₂/PVA nanofibre composites produced cylindrical fibres. The deposition of TiO₂ nano-particle on the fibres was clearly observed and was confirmed with the EDX mapping. At 15kV, the resultant fibre diameter was found to increase from 153 ± 23 nm to 191 ± 26 nm when the TiO₂ concentration increased from 1wt% to 10wt%. In addition, the membrane area of TiO₂/PVA nanofibre composites decrease with increasing of applied voltage. From the experimental works, it shows the feasibility to form TiO₂/PVA nanofibre composites using electrospinning. Further research will be focused on the potential applications of the TiO₂/PVA nanofibres.

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References

- [1] Hashimoto K, Irie H & Fujishima A (2005), TiO₂ Photocatalysis: A historical overview and future prospects. *Japanese Journal of Applied Physics* 44, 8269-8285.
- [2] Gupta KK, Jassal M & Agrawal A (2008), Sol-gel derived titanium dioxide finishing of cotton fabric for self cleaning. *Indian Journal of Fiber and Textile Research*, 33 (4), 433-450.
- [3] Veronovski N, Rudolf A, Smole MS & Gersak J (2009), Self-cleaning and handle properties of TiO₂-modified textiles. *Fibers and polymers*, 10(4), 551-556.
- [4] Razzaz A, Ghorban S, Hosayni L, Irani M & Aliabadi M (2016), Chitosan nanofibers functionalized by TiO₂ nanoparticles for the removal of heavy metal ions. *Journal of the Taiwan Institute of Chemicals Engineers*, 58, 333-342.
- [5] Bozzi A, Yuranova T, Guasaquillo I, Laub D & Kiwi J (2005), Self-cleaning of modified cotton textile by TiO₂ at low temperature under daylight irradiation. *Journal of Photochemistry and Photobiology A: Chemistry*, 174 (2), 156-164.
- [6] Nasikhudin, Ismaya EP, Diantoro M, Kusumaatmaja A & Triyana K (2017), Preparation of PVA/TiO₂ composites nanofibers by using electrospinning method for photocatalytic degradation. *IOP conference series: Materials Science and Engineering*, 202, 1-6.
- [7] Albetran H, Dong. Y & Low I.M (2015), Characterization and optimization of electrospun TiO₂/PVP nanofibers using Taguchi design of experiment method. *Journal of Asian Ceramic Societies*, 3(3), 292-300.
- [8] Mirjalili M & Zohoori S (2016), Review for publication of electrospinning and electrospun nanofibers technology in textile industry. *Journal of Nanostructure in Chemistry*, 6(3), 207-213.
- [9] Affandi NDN, Ahmad MR, Baharudin A & Abdullah@Shukry NA (2012), Effect of crosslinking on the solubility and morphological structures of the PVA nanofibres. *Humanities, Science and Engineering (CHUSER), 2012 IEEE Colloquium on*, 458-462.
- [10] Zhang C, Yuan X, Wu L, Han Y & Sheng J (2005), Study on morphology of electrospun poly(vinyl) alcohol mats. *European Polymer Journal*, 41(3), 423-432.
- [11] Affandi NDN (2010). Studies on fabrication, morphology and performances of bi-layer electrospun nanofibre membranes (Doctoral dissertation). Retrieved from RMIT University http://researchbank.rmit.edu.au/view/rmit:10383/Nor_Affandi.pdf
- [12] Khajavi R & Damerchely R (2007), Effect of Polyvinyl Alcohol concentration in Spinning Dope on Diameter, Beads and HHS of produced nanofibers. *Biological Sciences*. 10(2), 314-317.
- [13] Supaphol P, Mit-uppatham C & Nithitanakul M (2005), Ultrafine electrospun polyamide 6 fibers: Effects of solvents system and emitting electrode polarity on morphology and average fiber diameter. *Macromolecular Materials and Engineering*. 290 (9), 933-942.